First principles calculations of the copper silicide precipitates R. Taniguchi*, S. R. Nishitani, Y. Ohno¹⁾ and I. Yonenaga¹⁾

Figure 1 shows that two Cu₃Si phases are both more stable than the

[1] N. E. Christensen., PHYSICAL REVIEW., 32 No.1,

Package) code.

Reference:

(1985), pp.209-210.

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1) Institute for Materials Research, Tohoku University, Katahira 2-1-1, Aoba-ku Sendai, 980-8577 Japan. Very recently, the authors have reported the experimental observation of the precipitates of Cu₃Si in a heavily

by the first principles calculations. We adopted two atomistic structure models of Cu₃Si based on the Zintl phases [1]; one is a bcc based structure (D03 type), and the other is a fcc based structure (L12 type). The reported stable intermetallic structure of Cu₇Si₂ (the η phase) is also calculated. The energy calculations with the outer and inner relaxes of atomistic structures are performed by the VASP(Vienna Ab-initio Simulation

-4.122

-4.144

-4.153

-4.186

Fig.1 Energy composition diagram of Cu-Si system.

copper doped Si. In this research, the energetic assessment of this precipitation behavior has been investigated

segregation limit of pure Cu and Si. Although the experimentally observed precipitates show the very close structure with the D03 model, the L12 model is more stable -3.5 Cu than the D03 model. Both models of Cu₃Si show no band gap, which indicates that the phases are conductor. The other phase of Cu₇Si₂ is also more stable than -4.5 the segregation limit. The tie line between pure Cu ∇ : η phase and Cu7Si2 drawn by the dashed line indicates that ▲: Segregation Limit the D03 model is metastable to Cu + Cu7Si2, and ♦ : D03 model the L12 model is most stable. Because the D03 o: L12 model -5.5 model shows the good lattice coherency with pure 50.0 100.0 Si, it would appear as the structure of the Si contents [%] precipitates.